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<b>(54) Title:</b> CRYSTALLINE TERIPARATIDE  <b>(57) Abstract</b>  Pure, stable crystalline forms of parathyroid hormone, particularly teriparatide, are described as well as methods of preparation and purification.		

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## CRYSTALLINE TERIPARATIDE

### TECHNICAL FIELD

This invention relates to a pure crystalline form of a parathyroid hormone.

- 5 More particularly, the invention relates to the crystalline form of teriparatide, PTH(1-34), and methods of preparation and purification of the fragmented hormone.

### BACKGROUND OF THE INVENTION

- Parathyroid hormone (PTH) is a secreted, 84 amino acid product of the  
10 mammalian parathyroid gland that controls serum calcium levels through its action on various tissues, including bone. Studies in humans with certain forms of PTH have demonstrated an anabolic effect on bone, and have prompted significant interest in its use for the treatment of osteoporosis and related bone disorders.

- Using the N-terminal 34 amino acids of the bovine and human hormone for  
15 example, which by all published accounts are deemed biologically equivalent to the full length hormone, it has been demonstrated in humans that parathyroid hormone enhances bone growth particularly when administered in pulsatile fashion by the sub-cutaneous and intravenous routes. A slightly different form of PTH, human PTH(1-38) has shown similar results.

- 20 PTH preparations have been reconstituted from fresh or lyophilized hormone, and incorporate various forms of carrier, excipient and vehicle. Most are prepared in water-based vehicles such as saline, or water acidified typically with acetic acid to solubilize the hormone. The majority of reported formulations also incorporate albumin as a stabilizer (see for example Reeve et al., Br. Med. J., 1980, 280:6228; Reeve et al., Lancet, 1976, 1:1035; Reeve et al., Calcif. Tissue Res., 1976, 25 21:469; Hodsman et al., Bone Miner, 1990, 9(2):137; Tsai et al., J. Clin. Endocrinol Metab., 1989, 69(5):1024; Isaac et al., Horm. Metab. Res., 1980, 12(9):487; Law et al., J. Clin Invest. 1983, 72(3):1106; and Hulter, J. Clin Hypertens, 1986, 2(4):360). Other reported formulations have incorporated an excipient such as mannitol, which  
30 is present either with the lyophilized hormone or in the reconstitution vehicle. Formulations representative of those employed for human studies include a human

PTH(1-34) (SEQ ID NO: 2) preparation consisting, upon reconstitution, of mannitol, heat inactivated human serum albumin, and caproic acid (a protease inhibitor) as absorption enhancer (see Reeve et al., 1976, *Calcif. Tissue Res.*, 21, Suppl., 469-477); a human PTH(1-38) preparation reconstituted into a saline vehicle (see  
5 Hodsmen et al., 1991, 14(1), 67-83); and a bovine PTH(1-34) preparation in aqueous vehicle pH adjusted with acetic acid and containing albumin. There is also an International Reference preparation which for human PTH (1-84) (SEQ ID NO: 1) consists of 100 ng of hormone ampouled with 250 µg human serum albumin and 1.25 mg lactose (1981), and for bovine PTH (1-84) consists of 10 µg lyophilized  
10 hormone in 0.01M acetic acid and 0.1% w/v mannitol (see Martindale, *The Extra Pharmacopoeia*, The Pharmaceutical Press, London, 29th Edition, 1989 at p. 1338).

Commercial exploitation of parathyroid hormone requires the development of a formulation that is acceptable in terms of storage stability and ease of preparation. Because it is a protein and thus far more labile than the traditionally  
15 small molecular weight drugs, however, the formulation of parathyroid hormone presents challenges not commonly encountered by the pharmaceutical industry. Furthermore, like other proteins that have been formulated successfully, PTH is particularly sensitive to oxidation, deamidation and hydrolysis and further requires that its N-terminal and C-terminal sequences remain intact in order to preserve  
20 bioactivity.

### **SUMMARY OF THE INVENTION**

The present invention provides methods of preparing crystalline forms of a fragmented parathyroid hormone (PTH) which heretofore have not been reported.  
25 The advantages of a crystalline form for the hormone are purity of the product and storage stability. Thus, for example, a crystalline form of PTH may be easily dissolved in a sterile solution in vials for parenteral administration. As a crystalline material, PTH may also be formulated, if desired, into other compositions such as, for example, tablets, capsules or suppositories.

30 Accordingly, in a first aspect the present invention is a novel crystalline form of a parathyroid hormone selected from the group consisting of PTH(1-34), PTH(1-

37), PTH(1-38), and PTH(1-41), and, in particular crystalline human PTH(1-34) (SEQ ID NO: 2), generically known as teriparatide, in the form of tetragonal plates or cubic crystals, both having a space group P422 and the following unit cell constants:  $a = b = 91.071 \text{ \AA}$ ,  $c = 37.665 \text{ \AA}$ ,  $\alpha = \beta = \gamma = 90^\circ$ .

5           A second aspect of the present invention is a process for purifying a parathyroid hormone including the steps of:

- (a)     providing an aqueous solution of said hormone at a concentration of about 5 to 40 mg per ml;
- (b)     mixing said solution with a reservoir solution comprising  
10       organic solvent at a concentration of about 5 to about 50 volume percent and a buffer at a concentration to maintain the pH between about 6.0 and about 12.0; and
- (c)     allowing the resulting solution to stand at a predetermined  
15       time at a temperature between about room temperature and about  $4^\circ\text{C}$  until plate-like crystals form.

A third aspect of the present invention is a process for purifying a parathyroid hormone where the steps include:

- (a)     providing an aqueous solution of said hormone at a concentration of about 5 to 40 mg/ml;
- (b)     mixing said solution with a reservoir solution comprising a  
20       phosphate or formate salt precipitant at the concentration of about 0.5M to 2.5M and, optionally, an aqueous buffer solution, or a mixture thereof, at a concentration to maintain a pH between about 4.0 and 6.0; and
- (c)     allowing the resulting solution to stand at a predetermined  
25       time at a temperature between about room temperature and about  $4^\circ\text{C}$  until cubic crystals form.

A fourth aspect of the present invention is a novel crystalline form of a parathyroid hormone selected from the group consisting of PTH(1-34), PTH(1-37), PTH(1-38), and PTH(1-41), and, in particular crystalline human PTH(1-34) (SEQ ID  
30 NO: 2) in the form of hexagonal crystals having a space group P622 and the following cell constants:  $a = b = 30.169 \text{ \AA}$ ,  $c = 110.597 \text{ \AA}$ ,  $\alpha = \beta = 90^\circ$ ,  $\gamma = 120^\circ$ .

Finally, as a fifth aspect, the present invention is a process for purifying a parathyroid hormone comprising:

- (a) providing a solution of a parathyroid hormone selected from the group consisting of PTH(1-34), PTH(1-37), PTH(1-38), and PTH(1-41) at a concentration of about 5 to about 40 mg/ml in about 10 to about 30 volume percent glycerol;
- (b) mixing said solution with a reservoir solution comprising organic solvent at a concentration of about 5 to about 50 volume percent, ammonium sulfate at a concentration of about 0.5M to 3.0M, and a buffer at a concentration to maintain the pH of the solution between about 3.0 and about 7.0; and
- (c) allowing the resulting solution to stand at a predetermined time at a temperature between about room temperature and about 4°C until hexagonal crystals form.

#### **BRIEF DESCRIPTION OF THE DRAWINGS**

FIGURE 1 is a photograph showing PTH(1-34) crystals grown from Example 1.

FIGURE 2 is a photograph showing PTH(1-34) crystals grown from Example 4.

#### **DETAILED DESCRIPTION**

The invention relates to a pure crystalline form of a fragmented parathyroid hormone in its most stable form for storage and direct use in pharmaceutical compositions for administering to human patients.

The crystalline material may incorporate as active ingredient fragments or variants of fragments of human PTH or of rat, porcine or bovine PTH that have human PTH activity as determined in the ovariectomized rat model of osteoporosis reported by Kimmel et al., Endocrinology, 1993, 32(4):1577.

The parathyroid hormone fragments desirably incorporate at least the first 34 N-terminal residues, such as PTH(1-34), PTH(1-37), PTH(1-38) and PTH(1-41). Alternatives in the form of PTH variants incorporate from 1 to 5 amino acid

substitutions that improve PTH stability and half-life, such as the replacement of methionine residues at positions 8 and/or 18 with leucine or other hydrophobic amino acid that improves PTH stability against oxidation and the replacement of amino acids in the 25-27 region with trypsin-insensitive amino acids such as histidine or other amino acid that improves PTH stability against protease. These forms of PTH are embraced by the term "parathyroid hormone" as used generically herein. The preferred hormone is human PTH(1-34) (SEQ ID NO: 2) also known as teriparatide. The hormones may be obtained by known recombinant or synthetic methods, such as described in U.S. Pat. No. 4,086,196, incorporated herein by reference.

PTH crystals embodying the invention may be prepared or grown from a reservoir or precipitant solution containing a buffer and, optionally, an organic solvent or salt, with the concentration of reagents and pH being carefully controlled within prescribed limits. Any of the well known basic crystallization techniques may be used for growth of the hormone crystals. Crystals may be grown, for example, at both room temperature and down to about 4°C, by using a sitting drop, hanging drop, seeding and/or batch crystallization apparatus. The batch method is obviously preferred for large scale preparations.

In the hanging drop method a small drop of protein solution is placed on a cover slip, or glass plate, which is inverted over a well of solution and sealed. The solution in the well contains a precipitating agent, which is also present in a lesser amount in the protein droplet. The function of the precipitating agent is twofold. First, the solution in the well is initially at a lower vapor pressure than the protein droplet so that evaporation progresses at a rate fixed by the difference in the vapor pressures and the distance by which the vapor (usually water) must diffuse. Secondly, the precipitating agent lowers the solubility of the protein in solution by competing with the protein for available solvent, and thus as evaporation from the protein droplet occurs the solution becomes supersaturated in protein. Under the appropriate conditions including pH, protein concentration and temperature, crystallization of the protein or macromolecule then occurs.

The batch methods generally involve the slow addition of a precipitating agent to an aqueous solution of protein until the solution just becomes turbid, at this point the container is sealed and left undisturbed for a predetermined time.

The PTH crystals may be obtained using either basic conditions or acidic  
5 conditions with any of the above methods.

Under basic conditions, the PTH fragment is first obtained in purified, lyophilized form. The lyophilized protein is dissolved in water at a concentration from about 5 mg/ml to about 40 mg/ml, preferably about 5 mg/ml to about 20 mg/ml. The aqueous protein solution is then mixed with a reservoir solution  
10 containing from about 5 to about 50 volume percent, preferably about 10 to 20 volume percent of an organic solvent and a buffer in an amount to maintain a pH range of about 6.0-12.0. The aqueous solution and the reservoir solution are preferably mixed in a 1:1 ratio.

The organic solvent used may be, for example and without limitation,  
15 methanol, ethanol, isopropanol, MPD, polyethylene glycol, ethylene glycol, or a mixture thereof. Any buffer may be used, preferably any biological buffer, e.g. the tris(hydroxymethyl)aminomethane salts, e.g. maleate or hydrochloride, or a glycine buffer. Alkali metal salts may also be used depending on the desired pH. For example, a citrate source such as sodium citrate may be used as a slightly basic  
20 buffer.

Crystallization in basic conditions affords plate-like crystals. Particularly, for example, teriparatide plate-like crystals have been grown using the above-described basic conditions within approximately 48 hours at room temperature and at 4°C with an approximate size of 0.1 mm x 0.1 mm x 0.05 mm.

25 In acid conditions, the PTH fragment is prepared in the same manner in an aqueous solution at a concentration of about 5 mg/ml to about 40 mg/ml, preferably about 5 mg/ml to about 20 mg/ml. The aqueous solution containing the protein is mixed with a reservoir solution containing a salt and/or buffer in an amount to maintain a pH between about 4.0 and about 6.0. The protein and buffer solutions are  
30 preferably mixed in a 1:1 ratio.



The reservoir solution contains a precipitant salt such as, for example, a phosphate formate at a concentration of about 0.5M to about 2.5M.

The buffer employed in the reservoir solution may be any biologically acceptable buffer at pH range 4.0-6.0. Such buffers are, for example, various  
5 phosphates, formates, acetates, tartrates and the like including mixtures thereof, which are in the form of their respective alkali metal salts, e.g. sodium and/or potassium. When a phosphate or formate salt is used as the precipitant salt, an additional buffer may be added only if desired.

Crystallization in acidic media affords cube-like crystals. Thus, for example,  
10 teriparatide cube-like crystals have been grown within approximately 48 hours using the above described acidic conditions at room temperature and at 4°C with an approximate size of 0.1 mm x 0.1 mm x 0.1 mm.

The preferred embodiment of the present invention is the crystalline form of teriparatide, human PTH(1-34) (SEQ ID NO: 2). Crystalline teriparatide prepared  
15 by the above-described methods has been characterized by x-ray crystallography as having a crystal system of tetragonal plates or cubic crystals with a symmetry or space group of P422 and the following cell dimension having unit cell constants:

$$a = b = 91.071 \text{ Angstroms } (\text{\AA}),$$

$$c = 37.665 \text{\AA},$$

20  $\alpha = \beta = \gamma = 90^\circ$

Under an alternate embodiment, the PTH fragment obtained as described above is dissolved in a solution containing from about 10 to about 30 volume percent glycerol, preferably about 20 volume percent, where the protein concentration is from about 5 mg to about 40 mg/ml, preferably about 5 mg/ml to  
25 about 2 mg/ml. The glycerol protein solution is then mixed with a reservoir solution containing from about 5 to about 50 volume percent, preferably about 5 to 20 volume percent of an organic solvent, and a buffer in an amount to maintain a pH range of about 3.0 to about 7.0. The protein solution and the reservoir solution are preferably mixed in a 1:1 ratio.

30 The organic solvents are those previously described. Isopropanol is preferred. Buffers previously described may also be used depending on the desired

pH within the 3-7 range. Preferred as a buffer is a citrate salt, e.g. sodium citrate. In addition to the above solvent and buffer, the reservoir solution contains a precipitant salt such as ammonium sulfate at a concentration of about 0.5 to 3.0M.

Crystallization in the glycerol media affords hexagonal-like crystals. Thus, for example, human PTH(1-34) (SEQ ID NO: 2) hexagonal-like crystals have been grown within about 12-24 hours using the above-described conditions at room temperature and at 4°C with an approximate size of 0.2 mm x 0.2 mm x 0.6 mm. X-ray crystallography of the hexagonal-like crystals show a crystal system with a symmetry or space group P622 and the following cell dimension having unit cell constants:  $a = b = 30.169\text{\AA}$ ,  $c = 110.597\text{\AA}$ ,  $\alpha = \beta = 90^\circ$ ,  $\gamma = 120^\circ$ .

The advantages of a crystalline form for fragmented PTH and especially for teriparatide is that the crystalline material once obtained provides the active therapeutic ingredient in its most pure and stable form. Thus, the crystalline material may be stored with a longer shelf life than prior forms of the hormone or solutions thereof. The crystalline material may then be directly used to prepare pharmaceutical formulations for administration to patients. Thus, for example, the crystalline PTH may be dissolved directly with sterile solutions into vials or cartridges for parenteral administration.

The PTH solution incorporates PTH in a medically effective amount, a term used with reference to amounts useful either therapeutically or in medical diagnosis. The particular amount of parathyroid hormone incorporated in the preparation can be pre-determined based on the type of PTH selected and on the intended end-use of the preparation. In one application, the preparations are exploited for therapeutic purposes, and particularly for the treatment of osteoporosis. Osteoporosis therapy entails administration of the reconstituted preparation by injection, desirably subcutaneous injection, in unit doses that reflect the prescribed treatment regimen but are, by way of example, for human PTH(1-34) (SEQ ID NO: 2), within the range from 25  $\mu\text{g}$  PTH/mL of injected solution to 1000  $\mu\text{g}$ /mL of injected solution per patient, with injection volumes being desirably from 0.02 to 1.3 mL. Accordingly, crystalline hPTH(1-34) (SEQ ID NO: 2) is desirably incorporated with a buffering

agent and excipient to form an aqueous solution containing PTH in a concentration range from 25 µg/mL to 1000 µg/mL, preferably 100 µg/mL to 500 µg/mL.

In one embodiment, the preparations are provided in a form that yields a unit container of 100-500 µg human PTH(1-34) (SEQ ID NO: 2) upon reconstitution into  
5 about 1 mL (0.8-1.2 mL) of the reconstitution vehicle, and the vials are accordingly loaded with about 1 mL of the aqueous PTH preparation, for subsequent freeze-drying.

Once the preparation is obtained as an aqueous solution containing desired amounts and concentrations of the buffering agent, excipient and PTH, individual  
10 vials are filled with the solution to the desired volume. The advantage of the present invention is that the above solution may be prepared with sterile water without the need to undergo a freeze-drying process.

In addition to their therapeutic use, the present PTH composition can be formulated and administered to aid in medical diagnosis, and particularly to assist in  
15 establishing the diagnosis of hypoparathyroidism and pseudohypoparathyroidism in hypocalcemic patients. Except for the dose of PTH, the composition of the PTH preparation will remain as described herein for therapeutic use. An intravenously infused, single dose of human PTH(1-34) (SEQ ID NO: 2) that is equal to 200 International Units of PTH activity is appropriate for this diagnostic purpose.  
20 Diagnosis is then made by determining the effect of administered PTH or urinary cAMP levels, with cAMP elevation being indicative of the hypoparathyroidism condition, rather than its pseudoform.

Since fragmented PTH is now available in a pure crystalline form, other dosage forms such as tablets, capsules, suppositories and the like may also be  
25 prepared for therapeutic or diagnostic administration.

The examples which follow are illustrative of the invention and are not intended to be limiting.

## EXAMPLES

### Example 1

Recombinant human rhPTH(1-34) (SEQ ID NO: 2) was prepared from material previously purified and lyophilized. The lyophilized protein was  
5 solubilized in water, at the concentration from 5 mg/ml to 20 mg/ml. The protein was mixed with reservoir solution (20% iso-propanol, 0.2M sodium citrate, 0.1M Tris.HCl buffer, pH=8.0) at 1:1 ratio on silicon coated glass cover slips. These cover slips were inverted and sealed over wells containing 1ml of reservoir solution. Crystals appeared as tetragonal plates within 48-96 hours with the size of 0.1 mm x  
10 0.1 mm x 0.05 mm. The product was confirmed by mass spectroscopy. A photograph of the crystals grown is shown in Fig. 1.

X-ray crystallography of the tetragonal plates showed a space group P422 and the following unit cell constants:  $a = b = 91.071\text{\AA}$ ,  $c = 37.665\text{\AA}$ ,  $\alpha = \beta = \gamma = 90^\circ$ .

### 15 Example 2

The same experiment was repeated as in Example 1 and mixed with a reservoir solution of 10% iso-propanol, 10% polyethylene glycol, 0.1M glycine buffer at pH=8.5.

### Example 3

20 Batch Crystallization: rhPTH (SEQ ID NO: 1) at a concentration of 10 mg/ml was mixed with 50% iso-propanol, 100mM Tris, pH=7.5, 0.2M sodium citrate at 1:1 ratio in the test tube. Crystals appeared within 48-96 hours with the size of 0.1mm x 0.1mm x 0.05 mm.

### Example 4

25 The protein was prepared in the same manner as in Example 1 and mixed in 1:1 ratio with reservoir solution (0.8M potassium phosphate, 0.8M sodium phosphate, pH=5.0). Cubic crystals appeared within 48 hours at room temperature with the size of 0.1mm x 0.1mm x 0.1mm. A photograph of the crystals grown is shown in Fig. 2.

30 X-ray crystallography of the cubic crystals showed a space group P422 and the following unit cell constants:  $a = b = 91.071\text{\AA}$ ,  $c = 37.665\text{\AA}$ ,  $\alpha = \beta = \gamma = 90^\circ$ .

Example 5

The protein was prepared the same way as Example 1 and mixed in 1:1 ratio with reservoir solution (0.8M sodium formate, 0.1M sodium acetate buffer at pH=4.5). The same crystals appeared as in Example 4.

5 Example 6

rhPTH(1-34) (SEQ ID NO: 2) was prepared from material previously purified and lyophilized. The lyophilized protein was solubilized in 20% glycerol, at the concentration from 5 mg/ml to 20 mg/ml. The protein was mixed with reservoir solution (2.4M ammonium sulfate, 5% isopropanol, 0.2 sodium citrate buffer, 10 pH=5.0) at 1:1 ratio on silicon coated glass cover slips. These cover slips were inverted and sealed over wells containing 1ml of reservoir solution. Hexagonal crystals appeared within 12-24 hours with the size of 0.2mm x 0.2mm x 0.6mm.

X-ray crystallography of the hexagonal crystals showed a space group P622 and the following unit cell constants:  $a = b = 30.169\text{\AA}$ ,  $c = 110.597\text{\AA}$ ,  $\alpha = \beta = 90^\circ$ ,  $\gamma$  15  $= 120^\circ$ .

## WE CLAIM:

1. A parathyroid hormone in crystalline form selected from the group consisting of PTH(1-34), PTH(1-37), PTH(1-38), and PTH(1-41).
2. The crystalline hormone of claim 1, said hormone being human PTH(1-34).
3. The crystalline hormone of claim 2, in the form of tetragonal plates having a space group P422 and the following unit cell constants:  $a = b = 91.071 \text{ \AA}$ ,  $c = 37.665 \text{ \AA}$ ,  $\alpha = \beta = \gamma = 90^\circ$ .
4. The crystalline hormone of claim 2 in cubic crystalline form having a space group P422 and the following unit cell constants:  $a = b = 91.071 \text{ \AA}$ ,  $c = 37.665 \text{ \AA}$ ,  $\alpha = \beta = \gamma = 90^\circ$ .
5. The crystalline hormone of claim 2, in hexagonal crystalline form having a space group P622 and the following unit cell constants:  $a = b = 30.169 \text{ \AA}$ ,  $c = 110.597 \text{ \AA}$ ,  $\alpha = \beta = 90^\circ$ ,  $\gamma = 120^\circ$ .
6. A process for purifying a parathyroid hormone comprising:
  - (a) providing an aqueous solution of a parathyroid hormone selected from the group consisting of PTH(1-34), PTH(1-37), PTH(1-38), and PTH(1-41) at a concentration of about 5 to about 40 mg per ml;
  - (b) mixing said solution with a reservoir solution comprising organic solvent at a concentration of about 5 to about 50 volume percent and a buffer at a concentration to maintain the pH of the solution between about 6.0 and about 12.0; and
  - (c) allowing the resulting solution to stand at a predetermined time at a temperature between about room temperature and about  $4^\circ\text{C}$  until plate-like crystals form.
7. The process of claim 6, wherein the hormone is human PTH(1-34) (SEQ ID NO: 2).
8. The process of claims 6, wherein the organic solvent is a member selected from the group consisting of methanol, ethanol, isopropanol, MPD, polyethylene glycol, ethylene glycol and a mixture thereof.

9. The process of claim 6, wherein the hormone concentration is from about 5 to about 20 mg/ml.
10. The process of claim 6, wherein the organic solvent is at a concentration of about 10 to 20 volume percent.
- 5 11. The process of claim 6, wherein the aqueous solution and the reservoir solution are mixed in a ratio of about 1:1.
12. A process for purifying a parathyroid hormone comprising:
- (a) providing an aqueous solution of a parathyroid hormone selected from the group consisting of PTH(1-34), PTH(1-37), PTH(1-38),  
10 and PTH(1-41) at a concentration of about 5 to about 40 mg/ml;
- (b) mixing said solution with a reservoir solution comprising a phosphate or formate precipitant salt at a concentration of about 0.5M to about 2.5M and, optionally, an aqueous buffer solution, or a mixture thereof, at a concentration to maintain a pH between about 4.0 and about 6.0; and
- 15 (c) allowing the resulting solution to stand at a predetermined time at a temperature between about room temperature and about 4°C until cubic crystals form.
13. The process of claim 12, wherein the hormone is human PTH(1-34) (SEQ ID NO: 2).
- 20 14. The process of claim 12, wherein the buffer is selected from the group consisting of phosphates, formates, acetates, tartrates and mixtures thereof.
15. The process of claim 12, wherein the hormone concentration is from about 5 to about 20 mg/ml.
16. The process of claim 12, wherein the aqueous hormone solution and  
25 the buffer solution are mixed at a ratio of about 1:1.
17. A process for purifying a parathyroid hormone comprising:
- (a) providing a solution of a parathyroid hormone selected from the group consisting of PTH(1-34), PTH(1-37), PTH(1-38), and PTH(1-41) at a concentration of about 5 to about 40 mg/ml in about 10 to about 30  
30 volume percent glycerol;

(b) mixing said solution with a reservoir solution comprising organic solvent at a concentration of about 5 to about 50 volume percent, ammonium sulfate at a concentration of about 0.5M to 3.0M, and a buffer at a concentration to maintain the pH of the solution between about 3.0 and about 7.0; and

(c) allowing the resulting solution to stand at a predetermined time at a temperature between about room temperature and about 4°C until hexagonal crystals form.

18. The process of claim 17, wherein the hormone is human PTH(1-34)  
10 (SEQ ID NO: 2).

19. The process of claim 17, wherein the organic solvent is a member selected from the group consisting of methanol, ethanol, isopropanol, MPD, polyethylene glycol, ethylene glycol and a mixture thereof.

20. The process of claim 17, wherein the hormone concentration is from  
15 about 5 to about 20 mg/ml.

21. The process of claim 17, wherein the buffer is a citrate.

22. The process of claim 17, wherein the hormone solution and the buffer solution are mixed at a ratio of about 1:1.



FIG. 1

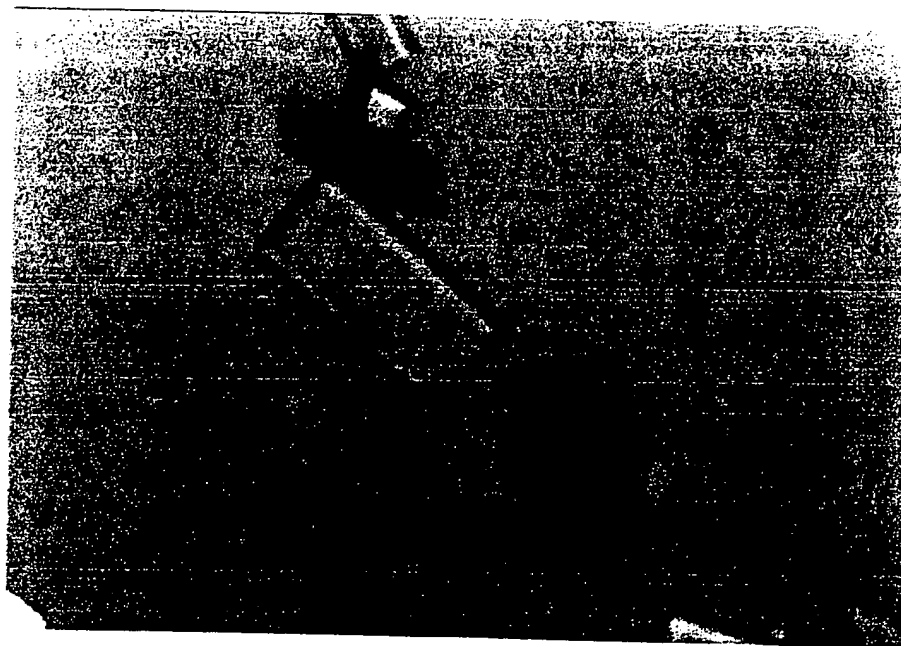
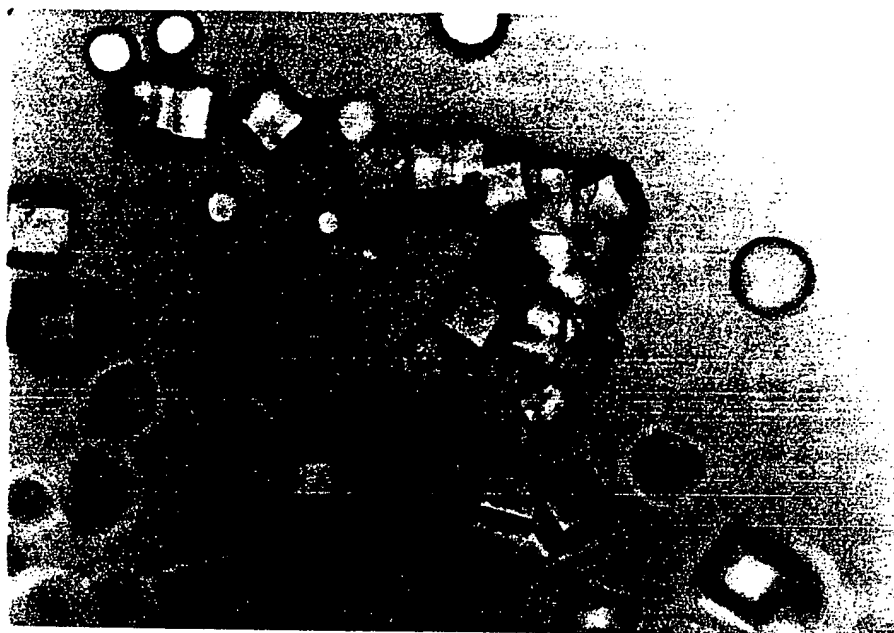


FIG. 2



## SEQUENCE LISTING

&lt;110&gt; Eli Lilly and Company

&lt;120&gt; CRYSTALLINE TERIPARATIDE

&lt;130&gt; 3797.17WO01

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&lt;141&gt; 1998-12-08

&lt;150&gt; 60/069,875

&lt;151&gt; 1997-12-18

&lt;160&gt; 2

&lt;170&gt; PatentIn Ver. 2.0

&lt;210&gt; 1

&lt;211&gt; 84

&lt;212&gt; PRT

&lt;213&gt; Homo sapiens

&lt;400&gt; 1

Ser	Val	Ser	Glu	Ile	Gln	Leu	Met	His	Asn	Leu	Gly	Lys	His	Leu	Asn
1				5					10					15	

Ser	Met	Glu	Arg	Val	Glu	Trp	Leu	Arg	Lys	Lys	Leu	Gln	Asp	Val	His
		20						25					30		

Asn	Phe	Val	Ala	Leu	Gly	Ala	Pro	Leu	Ala	Pro	Arg	Asp	Ala	Gly	Ser
	35						40					45			

Gln	Arg	Pro	Arg	Lys	Lys	Glu	Asp	Asn	Val	Leu	Val	Glu	Ser	His	Glu
50						55					60				

Lys	Ser	Leu	Gly	Glu	Ala	Asp	Lys	Ala	Asn	Val	Asp	Val	Leu	Thr	Lys
65					70					75					80

Ala Lys Ser Gln

&lt;210&gt; 2

&lt;211&gt; 34

&lt;212&gt; PRT

&lt;213&gt; Homo sapiens

&lt;400&gt; 2

Ser Val Ser Glu Ile Gln Leu Met His Asn Leu Gly Lys His Leu Asn  
1 5 10 15

Ser Met Glu Arg Val Glu Trp Leu Arg Lys Lys Leu Gln Asp Val His  
20 25 30

Asn Phe

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 98/26044

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C07K14/635

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C07K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	EP 0 748 817 A (TAKEDA CHEMICAL INDUSTRIES LTD) 18 December 1996 see page 1 - page 2, line 46 ---	1-22
Y	EP 0 553 539 A (SCHERING CORP) 4 August 1993 see page 1 - page 11, paragraph 3 ---	1-22
Y	US 5 419 278 A (CARTER DANIEL C) 30 May 1995 see column 1 - column 4 ---	1-22
Y	US 5 400 741 A (DETTA GEORGE T ET AL) 28 March 1995 see column 13 ---	1-22
Y	EP 0 225 050 A (FUJITSU LTD) 10 June 1987 see page 1 - page 10; table 1 -----	1-22



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

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Date of the actual completion of the international search

23 April 1999

Date of mailing of the international search report

04/05/1999

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# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

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